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Effect of Pr filling on the unfilled skutterudite CoSb₃

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Effect of rare earth elelements (RE) filling on unfilled skutterudite CoSb₃ is a interesting subject from two view points; i.e., application as thermoelctric materials and basic scientific interest. There have been reported quite a few trials to attack this subject. To increase the filling factor, some trials on partial replacement of Co site by Fe, $Ce_x(Fe_{1-y}Co_y)_4Sb_{12}$, have been reported, however 100 % filling was unsuccessful [1]. Since CoSb₃ is stable, it is difficult to increase the filling factor of RE in the antimony cage at ambient pressures. In contrast with the $REFe_4P_{12}$ and $RERu_4Sb_{12}$, no successful reports on 100% filling has been made even in $REFe_4Sb_{12}$ at ambient pressure [2]. Recently, we have succeeded in synthesizing high quality sample of $PrFe_4Sb_{12}$ by using high pressure synthesis, and found the physical properties are quite sensitive to the Pr site filling [3]. The magnetic ground state changes from a ferrimagnet to a highly enhanced paramagnet, which is consistent with the large density of states near the Fermi energy associated with Fe-3d electrons in the band calculation on $LaFe_4Sb_{12}$ [4]. This fact suggests an important role of 3d electrons on physical properties, we are trying to synthesize $PrCo_4Sb_{12}$ under high pressures, in which Fe is substituted for Co having one more d electron.

After some trials under defferent conditions, we have obtained the filled skutterudite compound $Pr_xCo_4Sb_{12}$. Figure 1 shows powder x-ray diffraction pattern of this sample. We cofirmed that the filled skutterudite was synthesized and a little unknown impurities included in the samples. The lattice constant was estimated to be 9.089 Å larger than that of unfilled skutterudite $CoSb_3$. In preliminary chemical composition analysis on field emission electron microscope, Pr site filling factor is estimated to be about 50 %. We will report the detail of synthesis techniques and the physical properties of $Pr_xCo_4Sb_{12}$ in this study.



Figure 1: Powder x-ray diffraction pattern of $Pr_xCo_4Sb_{12}$

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