(P1-22)

High pressure synthesis and transport properties of new filled skutterudite compounds RFe_4As_{12}

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Recent improvement of systematic studies for RT_4X_{12} (R; rare earth, T; Fe, Ru, Os, X; P, As, Sb), the various novel physical properties such as superconductors, heavy fermions, and metal-insulator transition are realized. Among them, only few filled skutterudite compounds including As are studied due to the difficulty of the making samples using flux method. Recently, we have succeeded in making new filled skutterudite compounds RFe_4As_{12} (R; light rare earth), using high pressure synthesizing method.

Polycrystalline RFe_4As_{12} were prepared at high temperatures and high pressures using KAWAI-type cubic-anvil high-pressure apparatus (UHP-1500) in Muroran-IT. The compounds were prepared by the reaction of stoichiometric amounts of each metal and arsenic powders at around 900 °C and 4.0 GPa. The samples were characterized by powder X-ray diffraction (RIGAKU RINT RAPID II) using Co K_{α} radiation and silicon as standard.



Figure 1: (a) Powder X-ray diffract pattern of $PrFe_4As_{12}$, and (b) rare earth dependence of lattice constant for RT_4As_{12} (R; rare earth, T; Fe, Ru, Os). Open circles show previous report and close circles show our results.

Figure 1 (a) shows the powder x-ray diffraction pattern of $PrFe_4As_{12}$. A similar pattern was also observed for the other compounds. All diffraction lines of compounds prepared at high pressure are assigned by the index of the filled skutterudite structure. The lattice constants of the compounds are determined as 8.2923 Å, 8.3080 Å, 8.3034 Å, and 8.3003 Å for $CeFe_4As_{12}$, $PrFe_4As_{12}$, $NdFe_4As_{12}$, and $SmFe_4As_{12}$. Then, we plot obtained lattice constants with other As-contained filled skutterudite compounds as shown in figure 1 (b). For $CeFe_4As_{12}$, obtained lattice constant is in good agreement with the previous report¹. The lattice constants systematically decrease due to the lanthanide contraction except Ce-based filled skutterudite compounds is also observed in other compounds such as $CeFe_4As_{12}$, suggesting the change of the valence.

[1] D.J. Braun and W. Jeitschko; J. Solid State Chem. **32** (1980) 357.